

RD-A138 146 PREPARATION OF FLUOROMETHYLENE OLEFINS(U) IOWA UNIV
IOWA CITY D J BURTON 28 DEC 83 ARO-16382. 16-CH
DARG29-79-C-0060

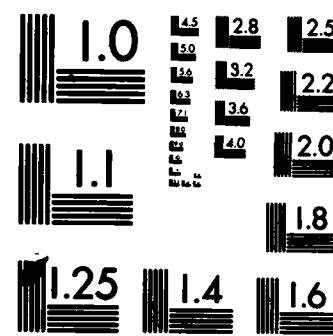
1/1

UNCLASSIFIED

F/G 11/9

NL





MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

AD A 1 3 8 1 4 6

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

DTIC FILE COPY

PREPARATION OF FLUOROMETHYLENE OLEFINS

FINAL REPORT

Donald J. Burton

December 28, 1983

U.S. Army Research Office

University of Iowa



Approved	for Public Release
1	2
3	4
5	6
7	8
9	10
11	12
13	14
15	16
17	18
19	20
21	22
23	24
25	26
27	28
29	30
31	32
33	34
35	36
37	38
39	40
41	42
43	44
45	46
47	48
49	50
51	52
53	54
55	56
57	58
59	60
61	62
63	64
65	66
67	68
69	70
71	72
73	74
75	76
77	78
79	80
81	82
83	84
85	86
87	88
89	90
91	92
93	94
95	96
97	98
99	100
101	102
103	104
105	106
107	108
109	110
111	112
113	114
115	116
117	118
119	120
121	122
123	124
125	126
127	128
129	130
131	132
133	134
135	136
137	138
139	140
141	142
143	144
145	146
147	148
149	150
151	152
153	154
155	156
157	158
159	160
161	162
163	164
165	166
167	168
169	170
171	172
173	174
175	176
177	178
179	180
181	182
183	184
185	186
187	188
189	190
191	192
193	194
195	196
197	198
199	200
201	202
203	204
205	206
207	208
209	210
211	212
213	214
215	216
217	218
219	220
221	222
223	224
225	226
227	228
229	230
231	232
233	234
235	236
237	238
239	240
241	242
243	244
245	246
247	248
249	250
251	252
253	254
255	256
257	258
259	260
261	262
263	264
265	266
267	268
269	270
271	272
273	274
275	276
277	278
279	280
281	282
283	284
285	286
287	288
289	290
291	292
293	294
295	296
297	298
299	300
301	302
303	304
305	306
307	308
309	310
311	312
313	314
315	316
317	318
319	320
321	322
323	324
325	326
327	328
329	330
331	332
333	334
335	336
337	338
339	340
341	342
343	344
345	346
347	348
349	350
351	352
353	354
355	356
357	358
359	360
361	362
363	364
365	366
367	368
369	370
371	372
373	374
375	376
377	378
379	380
381	382
383	384
385	386
387	388
389	390
391	392
393	394
395	396
397	398
399	400
401	402
403	404
405	406
407	408
409	410
411	412
413	414
415	416
417	418
419	420
421	422
423	424
425	426
427	428
429	430
431	432
433	434
435	436
437	438
439	440
441	442
443	444
445	446
447	448
449	450
451	452
453	454
455	456
457	458
459	460
461	462
463	464
465	466
467	468
469	470
471	472
473	474
475	476
477	478
479	480
481	482
483	484
485	486
487	488
489	490
491	492
493	494
495	496
497	498
499	500
501	502
503	504
505	506
507	508
509	510
511	512
513	514
515	516
517	518
519	520
521	522
523	524
525	526
527	528
529	530
531	532
533	534
535	536
537	538
539	540
541	542
543	544
545	546
547	548
549	550
551	552
553	554
555	556
557	558
559	560
561	562
563	564
565	566
567	568
569	570
571	572
573	574
575	576
577	578
579	580
581	582
583	584
585	586
587	588
589	590
591	592
593	594
595	596
597	598
599	600
601	602
603	604
605	606
607	608
609	610
611	612
613	614
615	616
617	618
619	620
621	622
623	624
625	626
627	628
629	630
631	632
633	634
635	636
637	638
639	640
641	642
643	644
645	646
647	648
649	650
651	652
653	654
655	656
657	658
659	660
661	662
663	664
665	666
667	668
669	670
671	672
673	674
675	676
677	678
679	680
681	682
683	684
685	686
687	688
689	690
691	692
693	694
695	696
697	698
699	700
701	702
703	704
705	706
707	708
709	710
711	712
713	714
715	716
717	718
719	720
721	722
723	724
725	726
727	728
729	730
731	732
733	734
735	736
737	738
739	740
741	742
743	744
745	746
747	748
749	750
751	752
753	754
755	756
757	758
759	760
761	762
763	764
765	766
767	768
769	770
771	772
773	774
775	776
777	778
779	780
781	782
783	784
785	786
787	788
789	790
791	792
793	794
795	796
797	798
799	800
801	802
803	804
805	806
807	808
809	810
811	812
813	814
815	816
817	818
819	820
821	822
823	824
825	826
827	828
829	830
831	832
833	834
835	836
837	838
839	840
841	842
843	844
845	846
847	848
849	850
851	852
853	854
855	856
857	858
859	860
861	862
863	864
865	866
867	868
869	870
871	872
873	874
875	876
877	878
879	880
881	882
883	884
885	886
887	888
889	890
891	892
893	894
895	896
897	898
899	900
901	902
903	904
905	906
907	908
909	910
911	912
913	914
915	916
917	918
919	920
921	922
923	924
925	926
927	928
929	930
931	932
933	934
935	936
937	938
939	940
941	942
943	944
945	946
947	948
949	950
951	952
953	954
955	956
957	958
959	960
961	962
963	964
965	966
967	968
969	970
971	972
973	974
975	976
977	978
979	980
981	982
983	984
985	986
987	988
989	989
991	992
993	994
995	996
997	998
999	999
1001	1002
1003	1004
1005	1006
1007	1008
1009	1009
1011	1012
1013	1014
1015	1016
1017	1018
1019	1019
1021	1022
1023	1024
1025	1026
1027	1028
1029	1029
1031	1032
1033	1034
1035	1036
1037	1038
1039	1039
1041	1042
1043	1044
1045	1046
1047	1048
1049	1049
1051	1052
1053	1054
1055	1056
1057	1058
1059	1059
1061	1062
1063	1064
1065	1066
1067	1068
1069	1069
1071	1072
1073	1074
1075	1076
1077	1078
1079	1079
1081	1082
1083	1084
1085	1086
1087	1087
1089	1089
1091	1092
1093	1094

Research Report:

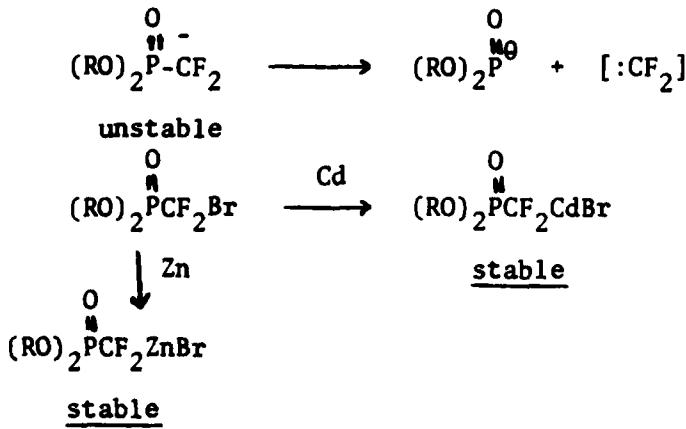
I. Statement of Problem:

The purposes of this investigation were: (1) to investigate the preparation of metal stabilized ylides and their utility in the preparation of fluoromethylene olefins, and (2) to investigate ylide-carbene reactions as a route to difluoromethylene olefins which avoids the limitations of the classical Wittig reaction.

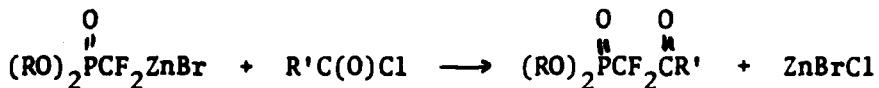
During the course of these investigations we discovered several new and novel reactions of difluorocarbene as well as a new and unique chain-extention reaction for the preparation of fluoroolefins and conjugated dienes.

II. Summary of Important Results:

A. Our work on the preparation of metal-stabilized ylides yielded four significant publications. We were able to prepare stable zinc and cadmium complexes of the difluoromethyl phosphonates ylide (cf. Publications #3 and 13). In contrast to the instability of difluoromethylene ylide (even at low temperatures) these metal stabilized ylides were thermally stable (Cd reagent to 60°C, Zn reagent to 100°C).

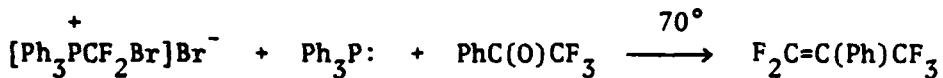


More importantly, the metal stabilized reagents mimic the unstable ylide in their chemical reactions and can be utilized in synthetic applications in which the free unstable ylide fails.



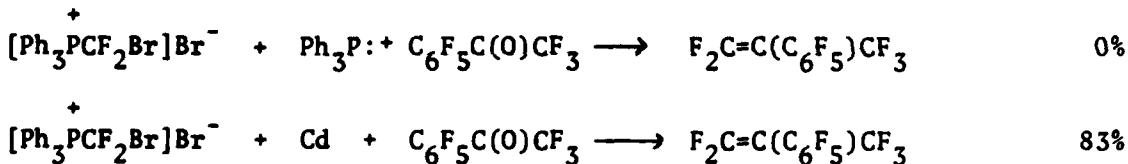
Cf. Publication #3

Similarly, the analogous phosphonium ylide $[R_3P-CF_2]$ is unstable and can only be generated and captured insitu via dehalogenation of bromodifluoromethyl-phosphonium salts with tertiary phosphines. However, if the olefin product is



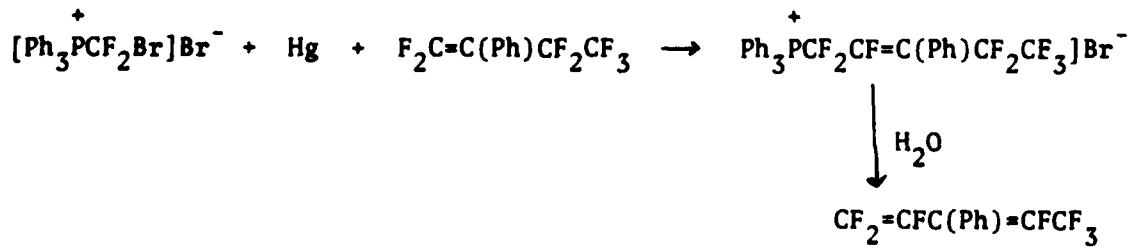
reactive towards the tertiary phosphine [utilized in generation of the ylide], the product is destroyed in a competitive side reaction with the tertiary phosphine.

However, we have no demonstrated that metal-stabilized ylides can circumvent this difficulty, and the preparation of highly reactive fluoroolefins is now possible via the use of these reagents.



Cf. Publication #12

Similar utility of metal dehalogenation of phosphonium salts was employed in the initial development of a chain-extension reaction of fluoroolefins.

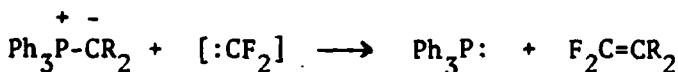


Cf. Publication #14

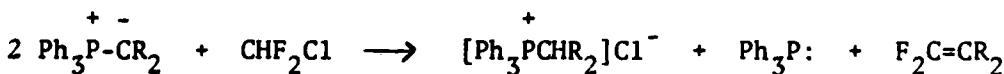
This novel reaction delineates the first unequivocal chain-extension method for the preparation of homologous fluoroolefins or dienes.

B. Our work on ylide-carbene chemistry has produced a novel route to difluoromethylene olefins in which all the phosphorus-containing moieties can be readily recycled. Thus, the major limitation (formation of Ph_3PO) of the classical Wittig reaction is avoided.

This method utilizes the ylide in a dual role - initially as a base to generate difluorocarbene insitu, and secondly as a nucleophilic trapping agent for the electrophilic carbene.



Overall Reaction: (sum of the two reactions above)



Note:

- 1) No Ph_3PO is produced.
- 2) The $[\text{Ph}_3\text{P-}^{\text{+}}\text{CHR}_2]\text{Cl}^- + \text{Ph}_3\text{P:}$ produced in the reaction can be readily recycled to give additional ylide. Thus, the expensive phosphorus reagents are not consumed.
- 3) Reaction is general and easily scaled up.
- 4) Cf. publication #1 for full details of this definitive paper.

C. The olefins obtained from the metal-stabilized ylide work and the ylide-carbene work provided us excellent model compounds to study the mechanism of the fluoride ion catalyzed isomerization of fluoroolefins. The mechanism of this process (concerted or carbanion) has been the subject of extensive

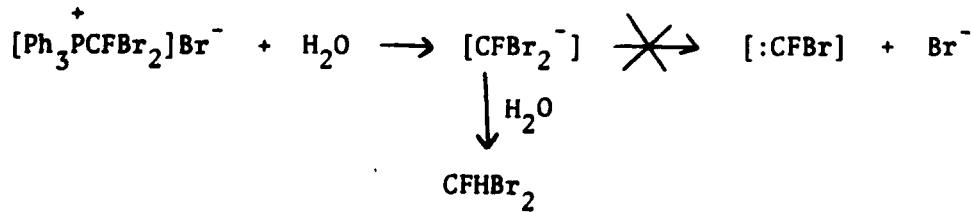
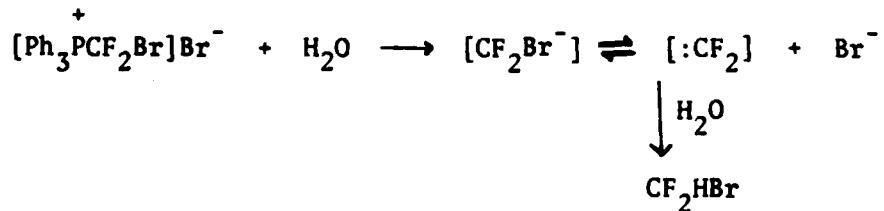


controversy. A detailed kinetic study of a series of 2-aryl-F-butenes gave unequivocal evidence in support of the carbanion mechanism. However, we did discover that there was a two step mechanism involved in the formation of the carbanionic intermediate, and a change in the rate limiting step was determined by the type of substituent present on the aryl ring.

This work provides the first unequivocal evidence for a carbanion intermediate in this type of catalytic reaction.

Full details are described in publication #11.

D. The mechanism of hydrolysis of fluorine-containing phosphonium salts was probed with the use of radioactive bromide. The mechanism (carbene or carbanion) depends upon the rate of collapse of the initially formed methide ion vs. the rate of protonation of the methide ion. For the bromodifluoromethyl phosphonium salts, unequivocal evidence was obtained for difluorocarbene as a transient intermediate. For dibromodifluoromethyl phosphonium salts, unequivocal evidence to support a carbanion intermediate was obtained. Thus, the mechanism of hydrolysis of these fluorinated phosphonium salts has been established as:



Publications #2 and 9 detail these results.

E. Othr carbene chemistry suggested by these results resulted in the work detailed in publications #9, #6, #7, and #4.

F. The work detailed in parts A and B led us to explore other routes to fluoro-olefins and ylides. These works are detailed in publications #15, #16, #10; and #5.

III. Publications:

List of Publications:

1. Ylide-Carbene Chemistry Synthesis of 1,1-Difluoro-1-Alkenes. Donald J. Burton and Gregory A. Wheaton, *J. Organic Chemistry*, 48, 917-927 (1983).
2. The Hydrolysis of Dibromofluoromethyl Triphenylphosphonium Bromide. Donald J. Burton, R.M. Flynn, R.G. Manning and R.M. Kessler, *J. Fluorine Chemistry*, 21, 371-376 (1982).
3. A Useful Zinc Reagent For The Preparation of 2-Oxo-1,1-Difluoroalkylphosphonate Donald J. Burton, Takashi Ishihara, and Masamichi Maruta, *Chemistry Letters*, 755-758 (1982).
4. Preparation, Stability, and Acidity of Difluoromethylene Bis-Phosphonic Acid. D.J. Burton, D.J. Pietrzyk, T. Ishihara, T. Fonong, and R.M. Flynn, *J. Fluorine Chemistry*, 20, 617-626 (1982).
5. A Convenient Preparation of Deuterated Fluoroolefins. Donald J. Burton and Francis J. Mettelle, *J. Fluorine Chemistry*, 20, 157-161 (1982).
6. Difluoromethylene Exchange In The Preparation of Fluorinated Bis-Phosphonates. Donald J. Burton and Richard M. Flynn, *J. Fluorine Chemistry*, 20, 121-126 (1982).
7. Preparation of Halo-F-Methanes Via Potassium Fluoride-Halogen Cleavage of Halo-F-Methyl Phosphonium Salts. D.J. Burton, S. Shin-ya and H.S. Kesling, *J. Fluorine Chemistry*, 20, 89-97 (1982).
8. Synthesis of Bromodifluoromethyl Phenyl Sulfide, Sulfoxide, and Sulfone. Donald J. Burton and Denise M. Wiemers, *J. Fluorine Chemistry*, 18, 573-582 (1981).
9. The Hydrolysis of Bromodifluoromethyl Triphenylphosphonium Bromide. R.M. Flynn, R.G. Manning, R.M. Kessler, D.J. Burton and S.W. Hansen, *J. Fluorine Chemistry*, 18, 525-531 (1981).
10. Tributylarsonium-2,2,3,3,4,4-hexafluorocyclobutane Ylide. Preparation and Cleavage. Donald J. Burton and Paul D. Vander Valk, *J. Fluorine Chemistry*, 18, 413-416 (1981).

11. Fluoride Ion Catalyzed Isomerization of 2-Aryl-F-Butenes. Donald J. Burton and James A. Headley, *J. Fluorine Chemistry*, 18, 323-356 (1981).
12. Metal Dehalogenation Route To Reactive Fluoroolefins. D.J. Burton, H.S. Kesling and D.G. Naae, *J. Fluorine Chemistry*, 18, 293-298 (1981).
13. Preparation, Stability, Reactivity, and Synthetic Utility of a Cadmium Stabilized Complex of Difluoromethylene Phosphonic Acid Ester. Donald J. Burton, Ryutaro Takei, and Seiji Shin-ya, *J. Fluorine Chemistry*, 18, 197-202 (1981).
14. Difluoromethylene Chain-Extension Reactions. Preparation of Fluorinated Alkenes and Alkadienes From Olefin Precursors. Donald J. Burton, Yoshio Inouye and James A. Headley, *J. Am. Chem. Soc.*, 102, 3980-3982 (1980).
15. The Preparation of Alpha, Halo, Beta, Beta-Difluorostyrenes. D.J. Burton, A.L. Anderson, R. Takei, H.F. Koch and T.L. Shih, *J. Fluorine Chemistry*, 16, 229-235 (1980).
16. Convenient Procedures For Conversion of Carbonyl Compounds To gem-Difluoroolefins and Their Selective Reductions To Monofluoroolefins. Sei-ichi Hayashi, Takeshi Nakai, Nobuo Ishikawa, Donald J. Burton, Douglas G. Naae, and H.S. Kesling, *Chemistry Letters*, 983-986 (1979).

IV. Participating Scientific Personnel

Professor Donald J. Burton	* G.S. Shaw
Dr. T. Ishihara	J. Yao
Dr. C. Buss	** D. Wiemers
Dr. Seiji Shin-ya	** S.W. Hansen
Dr. Allan Bailey	

* received Ph.D. degree

** will receive Ph.D. degree in 1984

V. Appendices:

Copies of each publication (16) that resulted from this project.

END

FILMED

3.80

DTIC